

Certification Boulevai

Roy Pelletier

Test Your Knowledge of Various Water-Related Topics

- 1. What typically happens to the pH of a pHneutral water sample when sulfuric acid is added for pH adjustment?
 - A. The pH increases.
 - B. The pH decreases.
 - C. Sulfuric acid does not affect Ph.
- 2. A water plant has a ground storage reservoir that is 100 feet in diameter and fills to its maximum operating depth of 23 feet in 3.75 hours. What is the average flow rate entering the tank in gpm?

A. 1,650 gpm	B. 7,136 gpm
C. 6,002 gpm	D. 8,546 gpm

3. If the finished water product temperature after treatment is 25 °C, what is the conversion to °F?
A. 19°F
B. 68°F
C. 103°F
D. 77°F

4.	What	typically	v happens	to the	pH of a pH-
1.	* * 11u t	ty preun,	mappens	to the	priorapri

neutral sample when alum is added? A. The pH increases. B. The pH decreases. C. Alum does not affect sample pH.

- 5. If the discharge head on an electrically driven centrifugal pump increases, what happens to the motor current?A. It remains the same. B. It goes up.C. It goes down. D. It will oscillate.
- 6. What is created when chlorine reacts with volatile organics?
 A. NH³
 B. THM
 C. CaCO³
 D. TMA
- 7. What is the volume of a tank if the flow entering is 500 gpm and the detention time is 1.2 hours?

A. 36,000 gals	B. 0.2344 mg
C. 69,399 gals	D. 24 mg

- 8. Which best describes dynamic head?
 - A. Pressure in a pumping system at rest.B. The total amount of pressure a pump
 - must overcome to properly discharge. C. Velocity of water in a main at full
 - pumping pressure.
 - D. Same as static head.
- 9. How many pounds of a chemical applied at the rate of 3 mg/L are required to dose 200,000 gallons?

SEND US YOUR QUESTIONS FOR CERTIFICATION BOULEVARD

Do you have a question or an exercise you would like to feature in "Certification Boulevard?" We'll be glad to publish it. Just send your question (with the answer) or your exercise (with the solution) to:

Roy A. Pelletier, Wastewater Consultant City of Orlando Public Works Department Environmental Services Wastewater Division 5100 L.B. McLeod Road, Orlando, FL 32811 roy.pelletier@cityoforlando.net Telephone 407-246-2213

There is no limit to the number of questions or exercises you may submit. Please include your name, city, and organization or company so we can give you credit.

A. 3 lbs	B. 8 lbs
C. 5 lbs	D. 10 lbs

- 10. What is the process called when chlorine is added to water until the chlorine demand has been satisfied?
 - A. Contact time
 - B. Hypo-chlorination
 - C. Reliquefaction
 - D. Breakpoint Chlorination

ANSWERS ON PAGE 74

Certification Boulevard Answer Key

From page 21

1. B. The pH decreases

Sulfuric acid is acidic (low pH), and will typically decrease the pH of a sample.

2. C. 6,002 gpm

Capacity of Tank at Max Level

- $=\pi r^2 x \text{ depth } x 7.48 \text{ gal/ft}^3$
- $= 3.14 x 50 ft. x 50 ft. x 23 ft. x 7.48 gal/ft^{3}$
- = 1,350,514 gals

Total Minutes of Pumping

- = 3.75 hrs x 60 mins/hr
- = 225 minutes

Average Flow Rate

- = Capacity, gals divided by Minutes Pumped
- *= 1,350,514 gals divided by 225 minutes*
- = 6,002 gpm

3. D. 77°F

 $^{\circ}C x 1.8 + 32 = ^{\circ}F$ 25°C x 1.8 + 32 = 77°F

4. B. The pH decreases

Alum (aluminum sulfate) is an acid and will typically reduce the pH of a sample after addition.

5. C. It goes down

Increasing the discharge pressure (head) on a centrifugal pump reduces the discharge flow; therefore, the pump does less work and the motor current goes down.

6. **B.** THM

Trihalomethane (THM) is created when chlorine reacts with volatile organics.

7. A. 36,000 gals

Tank Volume, mg

- = flow, $mgd \div 24$ hrs/day x detention time, hours
- $= (500 \text{ gpm } x \text{ 1440 mins/day} \div 1,000,000)$
- = 0.72 mgd
- $= 0.72 mgd \div 24 hrs/day x 1.2 hrs D.T.$
- = 0.036 mg
- Tank Volume, gals
- $= 0.036 mg \ x \ 1,000,000$
- = 36,000 gals

8. C. Velocity of water in a main at full pumping pressure

Dynamic head describes the velocity of a pumping system at full operating pressure. Dynamic head is fluid in motion ... static head is fluid at rest.

9. C. 5 lbs

Dose Rate, lbs

- = volume, mg x conc., ppm x 8.34
- = (200,000 gals ÷ 1,000,000) x 3 mg/L x 8.34 lbs/gal
- = 0.2 mg x 3 ppm x 8.34 lbs/gal
- = 5.0 lbs

10. D. Breakpoint Chlorination

Breakpoint chlorination is accomplished after all the demand for chlorine has been satisfied and free chlorine residual is achieved.